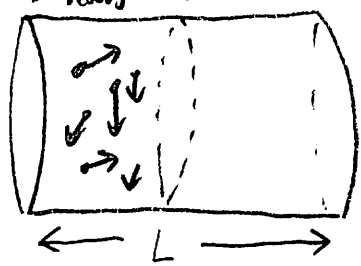


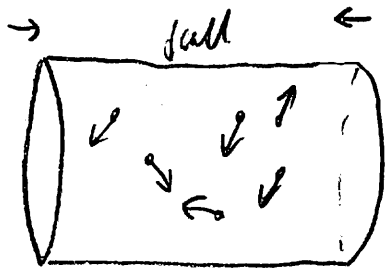
# More on the approach to equilibrium

n molecules  
kinetic gas  
→ half ←



positions in 0-L/2  
momenta in 0-P

spontaneously expands  
→



positions in 0-L  
momenta in 0-P

same since  
no work  
extracted

Volume in  
phase space

$$(P L/2)^{3n}$$

HUGE INCREASE  
Factor  $(1/2)^{3n}$  ↑ BIG!

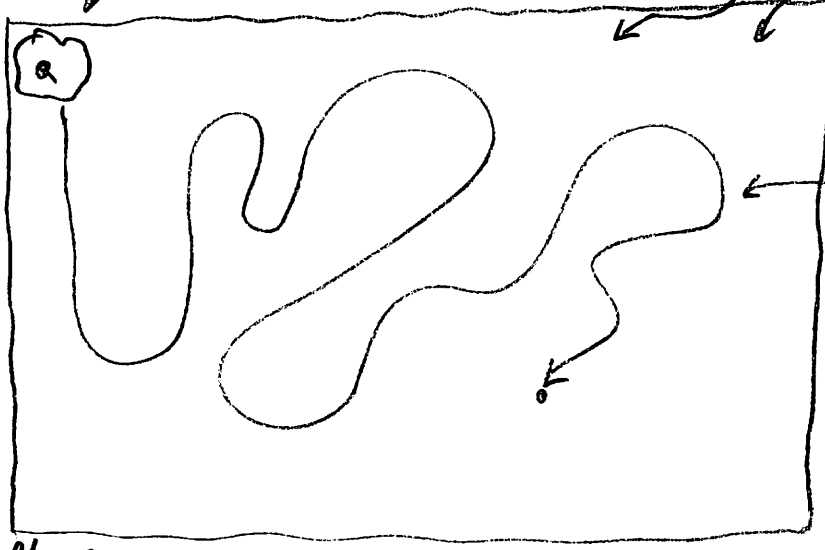
Volume in  
phase space

$$(P L)^{3n}$$

compressed gas

uncompressed gas

phase point starts here



Very likely to go here and stay here.

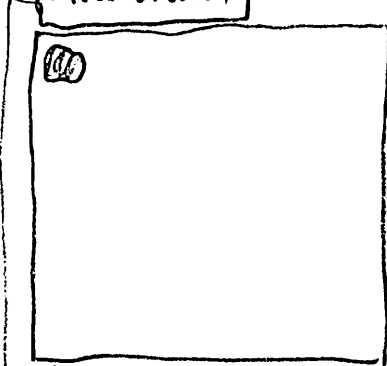
phase space

Poincaré recurrence  
... but will eventually go back to phase volume of compressed gas

# Fine & Coarse Graining

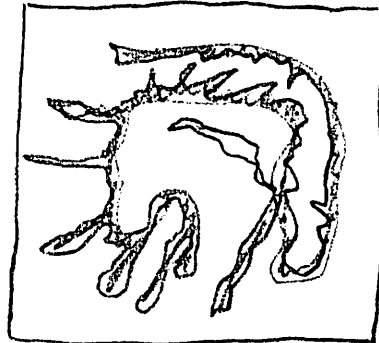
Expansion of the gas again:

Fine Grain



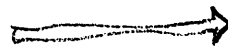
phase space

Liouville theorem  
 →  
 phase volume preserved



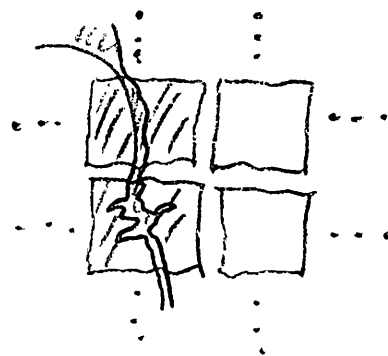
same volume stretched out into spidery filaments

Entropy =  $k \ln$  phase volume

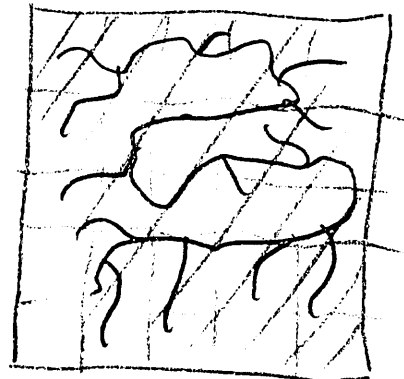
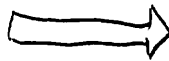
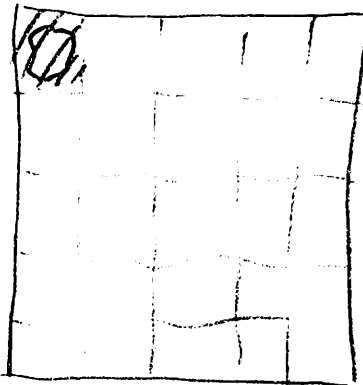


NO INCREASE IN ENTROPY

"Coarse grain" = Divide phase space into cells. Cell is occupied if any point in it accepts migrating phase volume



Coarse Grain



Entropy =  $k \ln$  volume  $(PL/2)^{3N}$

— increase by  $k 3N \ln 2$  →

Entropy =  $k \ln$  volume  $(PL)^{3N}$