

HW #6 Chem 1410

① Express in atomic units

a) the energy of the $1s \rightarrow 2p$ transition of H.

$$E = -\frac{1}{2n^2} \Rightarrow E(1s) = -\frac{1}{2}, E(2p) = -\frac{1}{8}$$

$$\Delta E = \frac{1}{2} - \frac{1}{8} = \frac{3}{8} \text{ a.u.}$$

b) the ground state energy of an electron in a box of width 5 \AA

$$E = \frac{h^2}{8ma^2} = \frac{4\pi^2 \hbar^2}{8ma^2} = \frac{\pi^2 \hbar^2}{2ma^2} = \frac{\pi^2}{2(1)[5/0.529]^2}$$

$$E = \frac{\pi^2}{2(9.45)^2} = 0.055 \text{ a.u.}$$

change to a.u.

Note: the conversion of the width "a" to a.u.

②

Expand out the Slater determinant for the ground state of the Li atom.

$$|1s \bar{1s} 2s| = \frac{1}{\sqrt{6}} \left\{ \begin{aligned} &1s \bar{1s} 2s - \bar{1s} 1s 2s - 1s 2s \bar{1s} - 2s \bar{1s} 1s \\ &+ 2s 1s \bar{1s} + \bar{1s} 2s 1s \end{aligned} \right\}$$

③

8-25

Consider the $1s^2 \rightarrow 1s2s$ excited states of the He atom

$$\phi_1 = |1s2s|, \phi_2 = |\bar{1s}\bar{2s}|, \phi_3 = |1s\bar{2s}|, \phi_4 = |\bar{1s}2s|$$

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Calculate the energies possible for a wavefunction of the form $\psi = c_1\phi_1 + c_2\phi_2 + c_3\phi_3 + c_4\phi_4$

$$\begin{vmatrix} H_{11} - ES_{11} & H_{12} - ES_{12} & H_{13} - ES_{13} & H_{14} - ES_{14} \\ H_{21} - ES_{21} & H_{22} - ES_{22} & H_{23} - ES_{23} & H_{24} - ES_{24} \\ H_{31} - ES_{31} & H_{32} - ES_{32} & H_{33} - ES_{33} & H_{34} - ES_{34} \\ H_{41} - ES_{41} & H_{42} - ES_{42} & H_{43} - ES_{43} & H_{44} - ES_{44} \end{vmatrix} = 0$$

$S_{ii} = 1$, $S_{ij} = 0$, $i \neq j$ by spin or spatial orbital orthogonality

$$\left. \begin{array}{l} H_{12} = 0 \\ H_{13} = 0 \\ H_{14} = 0 \\ H_{23} = 0 \\ H_{24} = 0 \end{array} \right\} \text{by spin orthogonality}$$

$$H_{ij} = H_{ji}$$

$$\begin{vmatrix} H_{11} - E & 0 & 0 & 0 \\ 0 & H_{22} - E & 0 & 0 \\ 0 & 0 & H_{33} - E & H_{34} \\ 0 & 0 & H_{34} & H_{44} - E \end{vmatrix} = 0 = (H_{11} - E)(H_{22} - E) \begin{vmatrix} H_{33} - E & H_{34} \\ H_{34} & H_{44} - E \end{vmatrix}$$

$$\begin{aligned} H_{11} &= \langle 1s2s || H || 1s2s \rangle \\ &= \frac{1}{2} \langle (1s2s - 2s1s) | H^0 + \frac{1}{r_{12}} | (1s2s - 2s1s) \rangle \\ &= \langle 1s2s | H^0 | 1s2s \rangle + \langle 1s2s | \frac{1}{r_{12}} | 1s2s \rangle - \langle 1s2s | \frac{1}{r_{12}} | 2s1s \rangle \\ &= E_0 + J - K \end{aligned}$$

Here we made use of:

$$\langle 1s2s | H^0 | 1s2s \rangle = \langle 2s1s | H^0 | 2s1s \rangle$$

$$\langle 1s2s | \frac{1}{r_{12}} | 1s2s \rangle = \langle 2s1s | \frac{1}{r_{12}} | 2s1s \rangle$$

$$\langle 1s2s | \frac{1}{r_{12}} | 2s1s \rangle = \langle 2s1s | \frac{1}{r_{12}} | 1s2s \rangle$$

Also spin is simply integrated out since both spins are α , and $\langle 1s2s | H^0 | 2s1s \rangle = 0$.

by symmetry $H_{22} = H_{11} = E_0 + J - K$

$$H_{33} = \frac{1}{2} \langle 1s\bar{2}s - \bar{2}s1s | H^0 + \frac{1}{r_{12}} | 1s\bar{2}s - \bar{2}s1s \rangle$$

$$= \langle 1s\bar{2}s | H^0 | 1s\bar{2}s \rangle + \langle 1s\bar{2}s | \frac{1}{r_{12}} | 1s\bar{2}s \rangle \rightarrow J$$

$$- \langle 1s\bar{2}s | \frac{1}{r_{12}} | \bar{2}s1s \rangle \rightarrow 0 \text{ by spin orthogonality}$$

$$= E_0 + J$$

By symmetry $H_{44} = H_{33} = E_0 + J$

$$H_{34} = \frac{1}{2} \langle 1s\bar{2}s - \bar{2}s1s | H^0 + \frac{1}{r_{12}} | 1s\bar{2}s - \bar{2}s1s \rangle$$

$$= \langle 1s\bar{2}s | \frac{1}{r_{12}} | \bar{2}s1s \rangle = -K$$

Here we used the fact that the integrals over H^0 are 0 by spin or spatial orbital orthogonality and $\langle 1s\bar{2}s | \frac{1}{r_{12}} | 1s\bar{2}s \rangle = 0$ by spin orthogonality

$$0 = H_{11} - E \rightarrow E = E_0 + J - K$$

$$0 = H_{22} - E \rightarrow E = E_0 + J - K$$

$$\begin{vmatrix} H_{33} - E & H_{34} \\ H_{34} & H_{44} - E \end{vmatrix} = 0 = \begin{vmatrix} E_0 + J - E & -K \\ -K & E_0 + J - E \end{vmatrix}$$

$$\Rightarrow (E_0 + J - E)^2 = K^2$$

$$E = E_0 + J - K \text{ and } E = E_0 + J + K$$

So we get three roots with energy $E_0 + J - K$ and one with energy $E_0 + J + K$. The former correspond to the three components of the triplet state and the latter to the singlet state.

Why is $J \oplus$?

$$J = \langle 1s2s | \frac{1}{r_{12}} | 1s2s \rangle = \int |\psi(r_1)|^2 \frac{1}{r_{12}} |\psi(r_2)|^2 d\vec{r}_1 d\vec{r}_2$$

This is just the classical interaction between two (negative) charge distributions, and, hence, is positive.

$$\text{Show that } E = E_0 + J + K \rightarrow \psi_3 = \frac{1}{\sqrt{2}}(\phi_3 - \phi_4)$$

$$E = E_0 + J - K \rightarrow \psi_4 = \frac{1}{\sqrt{2}}(\phi_3 + \phi_4)$$

From the system of linear eq's.

$$(E_0 + J - E)c_3 - Kc_4 = 0$$

$$(E_0 + J - E_0 - J - K)c_3 - Kc_4 = 0 \Rightarrow -Kc_3 - Kc_4 = 0$$

$$\Rightarrow c_3 = -c_4 \Rightarrow \psi_3 = (\phi_3 - \phi_4) \rightarrow \frac{1}{\sqrt{2}}(\phi_3 - \phi_4) \in \text{Normalized}$$

$$\text{if } E = E_0 + J - K$$

$$K C_3 - K C_4 = 0 \Rightarrow C_3 = C_4$$

$$\psi_4 = (\psi_3 + \psi_4) \rightarrow \frac{1}{\sqrt{2}}(\psi_3 + \psi_4) \leftarrow \text{Normalized}$$

Expand ψ_3

$$\begin{aligned} \psi_3 &= \frac{1}{\sqrt{2}} \{ \psi_3 - \psi_4 \} = \frac{1}{\sqrt{2}} \left\{ \frac{1}{\sqrt{2}} (1s\bar{2}s - \bar{2}s1s) - (\bar{2}s2s - 2s\bar{1}s) \frac{1}{\sqrt{2}} \right\} \\ &= \frac{1}{2} \{ 1s2s\alpha\beta - 2s1s\beta\alpha - 1s2s\beta\alpha + 2s1s\alpha\beta \} \\ &= \frac{1}{2} \{ 1s2s(\alpha\beta + \beta\alpha) + 2s1s(\alpha\beta - \beta\alpha) \} \\ &= \frac{1}{2} (1s2s + 2s1s)(\alpha\beta - \beta\alpha) \end{aligned}$$

Expand ψ_4

$$\begin{aligned} \psi_4 &= \frac{1}{\sqrt{2}} \{ \psi_3 + \psi_4 \} = \frac{1}{2} \{ (1s\bar{2}s - \bar{2}s1s) + (1s2s - 2s\bar{1}s) \} \\ &= \frac{1}{2} \{ 1s2s\alpha\beta - 2s1s\beta\alpha + 1s2s\beta\alpha - 2s1s\alpha\beta \} \\ &= \frac{1}{2} \{ (1s2s)(\alpha\beta + \beta\alpha) - (2s1s)(\alpha\beta + \beta\alpha) \} \\ &= \frac{1}{2} (1s2s - 2s1s)(\alpha\beta + \beta\alpha) \end{aligned}$$

From above

$$^1E = E_0 + J + K$$

$$^3E = E_0 + J - K$$

$$J > 0 \text{ and } K > 0$$

So the triplet is lower in energy

$$J = \frac{34}{81}, \quad K = \frac{32}{(27)^2}$$

$$E_{GS} = -11/4 = \text{energy of ground state}$$

$$E_{1s} = -\frac{1}{2} \frac{2^2}{1^2} = -2$$

$$E_{2s} = -\frac{1}{2} \left(\frac{2^2}{2^2} \right) = -\frac{1}{2}$$

$$E_{GS} = -4 + 2 \left(\frac{5}{8} \right) = -11/4$$

\uparrow $2e^-$ in 1s \nwarrow from 1st order PT.

$${}^1E = E_0 + J + K$$

$$= -2 - \frac{1}{2} + \frac{34}{81} + \frac{32}{(27)^2} = -2.04 \text{ a.u.}$$

\uparrow \uparrow
 1s 2s

$${}^3E = E_0 + J - K$$

$$= -2 - \frac{1}{2} + \frac{34}{81} - \frac{32}{(27)^2} = -2.12 \text{ a.u.}$$

$$E_{GS} \rightarrow {}^1E = 2.75 - 2.04 = 0.71 \text{ a.u.} = 19.3 \text{ eV}$$

$$E_{GS} \rightarrow {}^3E = 2.75 - 2.12 = 0.63 \text{ a.u.} = 17.1 \text{ eV}$$

8-29.

(4)

What is the # of sets of m_l and m_s for nd^8

$$\frac{G!}{N!(G-N)!} = \frac{10!}{8!2!} = \frac{10 \cdot 9}{2} = 45 \text{ sets}$$

1S	1D	3P	3F	1G
1×1	1×5	3×3	3×7	1×9
1	+ 5	+ 9	+ 21	+ 9 = 45

⑤
8-32

$$nd^2 \rightarrow {}^1S, {}^1D, {}^1G, {}^3P, {}^3F$$

What are the possible J states?

$${}^1S \rightarrow J=0$$

$${}^1D \rightarrow J=2$$

$${}^1G \rightarrow J=4$$

$${}^3P \rightarrow J=2, 1, 0$$

$${}^3F \rightarrow J=4, 3, 2$$

Expect 3F_2 to be lowest in energy based on Hund's rules.

⑥

8-47 $ns^2 np^2 \rightarrow {}^2P_{1/2}, {}^2P_{3/2}$

Which is the ground state?

${}^2P_{3/2}$ by Hund's rules

$$E({}^2P_{1/2}) - E({}^2P_{3/2}) > \text{ along the sequence } F, \text{ Ce, Br, I.}$$

Suggests that the spin-orbit interaction grows with Z at #. (see problem 8-46.)