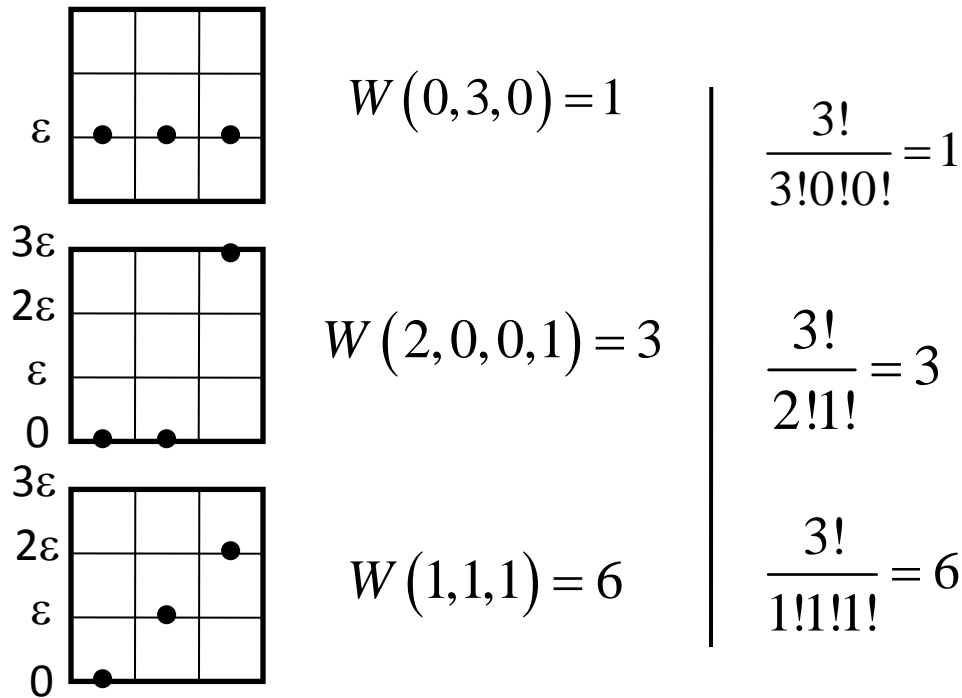


A simple example illustrating distributions, etc. in the Canonical ensemble

Consider each ensemble having one particle which can have energies of $0, \varepsilon, 2\varepsilon, 3\varepsilon$, etc.

Suppose there are 3 ensemble members
and $\langle E \rangle = \varepsilon$



most probably weight: prob. = 6/10

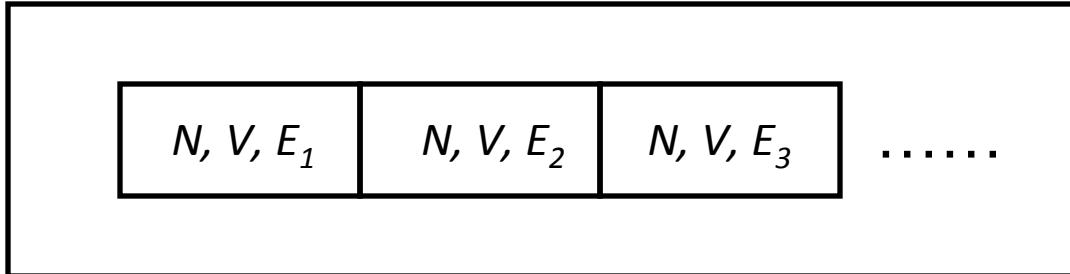
20 member ensemble

- most prob. Distribution $\langle E \rangle = \varepsilon$ with occurs 9.8×10^9 ways
- next most prob. member of distribution occurs 3.7×10^8 ways

10^{23} member ensemble

Distribution very strongly peaked

NVT (Canonical)



E_1, E_2, E_3 can differ

M ensembles

MV volume

MN particles

Energy E_1 can appear $\Omega(E_1)$ times

State	1,	2,	3, ...
Energy	E_1	E_2	E_3 ...
Occ #	a_1	a_2	a_3 ...

← numbers of systems with that state

$$\sum a_j = M$$

entire ensemble

$$\sum a_j E_j = E$$

has fixed E

All states consistent with constraints are equally probable

$$W(\mathbf{a}) = \frac{M!}{a_1! a_2! \dots}$$

of ways a particular distribution of a_j s can be achieved (assumes objects are distinguishable)

$$P_j = \frac{\bar{a}_j}{M} = \frac{1}{M} \frac{\sum_a W(\mathbf{a}) a_j(\mathbf{a})}{\sum_a W(\mathbf{a})} = \text{fraction of systems in } j^{\text{th}} \text{ energy state}$$

when M and a_j s are large, the spread in W is small and we can choose the set a_j^* which maximizes $W(a)$ (under constraints)

$$\begin{array}{l}
 P_j = \frac{a_j^*}{M} \\
 a_j^* = e^{-\alpha'} e^{-\beta E_j}, \quad \alpha' = \alpha + 1 \\
 P_j = \frac{e^{-\beta E_j(N,V)}}{\sum_j e^{-\beta E_j(N,V)}}
 \end{array}
 \quad \left| \quad
 \frac{\partial}{\partial a_j} \left[\ln W(a) - \alpha \sum a_k - \beta \sum a_k E_k \right] = 0
 \right.$$

$$Q(N, V, \beta) = \sum_j e^{-\beta E_j(N, V)} \quad \beta = \frac{1}{k_B T}$$

$$\left. \begin{array}{l}
 \langle E \rangle \leftrightarrow E \\
 \langle P \rangle \leftrightarrow P
 \end{array} \right] \text{ association with thermodynamics}$$

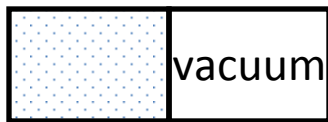
$$d\langle E \rangle = \underbrace{\sum E_j dP_j}_{\text{heat flow}} + \underbrace{\sum P_j dE_j}_{\text{work done on system}}$$

$$= \delta q_{rev} - \delta \omega_{rev}$$

change in populations
without changes in
energy levels

change in energies
without changes in
populations

$$Q = \underbrace{\sum_j e^{-E_j/kT}}_{\text{sum over states}} = \underbrace{\sum_E \Omega(N, V, E) e^{-E/kT}}_{\text{sum over levels}}$$



$V/2$



V

isolated
system

quantum states grows
E fixed

For an isothermal process (system connected to heat bath)
 remove constraint, # of accessible levels for a given E cannot <

$\Omega_2(NVE) > \Omega_1(NVE) \Rightarrow \Delta A < 0$ for spontaneous process

$$A = -kT \ln Q \quad \left| \quad \Delta A = -k_B T \ln \frac{Q_2}{Q_1} < 0$$

$$S = k \ln \sum e^{-E_j/kT} + \frac{1}{T} \frac{\sum E_j e^{-E_j/kT}}{\sum e^{-E_j/kT}} \xrightarrow{T \rightarrow 0} S = k \ln n, \quad \text{where } n \text{ is the degeneracy of the lowest level}$$

This is small compared to typical finite T values of $S (\propto Nk)$

In general, we choose the constant in the expression for S so that
 $S \rightarrow 0$ as $T \rightarrow 0$.

Can you think of a system where $S \neq 0$ in the $T \rightarrow 0$ limit?

Different ensembles

In general, in the $N \rightarrow \infty$ limit, it does not make a significant difference which ensemble we adopt

grand canonical (V, T, μ) E, N can fluctuate

$$\Xi(V, T, \mu) = \sum_N Q(N, V, T) e^{\mu N / kT}$$

N, P, T ensemble

$$\Delta(N, P, T) = \sum_E \sum_V \Omega(N, V, E) e^{-E/kT} e^{-PV/kT} \quad \text{partition function}$$

$$G = -kT \ln \Delta$$

$$S = k \ln \Omega(N, V, E)$$

micro canonical

$$Q(N, V, \beta) = \sum_E e^{-\beta E_j} = \sum_E \Omega(N, V, E) e^{-E/kT}$$
$$A = -kT \ln Q$$

$N V T$
canonical

$$\Xi = \sum_N e^{-\beta E_j + \beta \mu N_j} = \sum_N Q(N, V, T) e^{\mu N/kT}$$

sum over j for fixed N

grand
canonical

$$pV = kT \ln [\Xi(V, T, \mu)]$$

Chapter 1 derives Ω for translational motion of non-interacting particles

$$S = Nk \ln \left[\left(\frac{2\pi mkT}{h^2} \right)^{3/2} \frac{Ve^{5/2}}{N} \right]$$

$$\frac{p}{T} = \left(\frac{\partial S}{\partial V} \right)_{N,E} \Rightarrow pV = NkT$$

(ideal gas law)